



PRESLHY: Final Plan WP4 15/10/18

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Experimental Programme (at HSL)

I ARORATORY

- WP3: Unignited releases focussing on dispersion / source term
- WP4: Ignition phenomena focussing on electrostatics
 / condensed phase initiation / rapid phase transition
- WP5: Combustion characteristics including semiconfined / congested regions

WP4



- The key objective is to examine electrostatic charging in liquefied/multiphase mixtures; and ignition sensitivity of energetic multiphase mixtures of hydrogen and oxygen
 - E4.3 Electrostatic measurement of cold plume (focussing on accidental spillage scenarios)
 - E4.5 Ignition of H2/condensed O2 phase





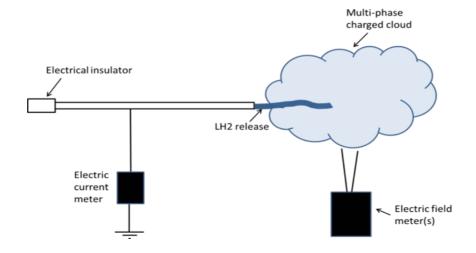
Work	Experimental	Test No.	Experiment Title	Release	Insulated Pipe	Field Measurement
Package	Subtask	lest No.		Orifice	Length	Method
4	4.3	4.3.1 (Test 3.5.3)	Electrostatic plume measurement	1/4"	1.0 m	Faraday cage
4	4.3	4.3.2 (Test 3.5.3)	Electrostatic plume measurement	1/4"	1.0 m	Field meter
4	4.3	4.3.3 (Test 3.5.1)	Electrostatic plume measurement	1"	1.0 m	Faraday cage
4	4.3	4.3.4 (Test 3.5.1)	Electrostatic plume measurement	1"	1.0 m	Field meter
4	4.3	4.3.5	Electrostatic plume measurement	1"	4.0 m	Faraday cage
4	4.3	4.3.6	Electrostatic plume measurement	1"	4.0 m	Field meter

Some release conditions will be as for WP3, 3.5.1 and 3.5.3

WP4 - Electrostatic plume measurement – Field meter



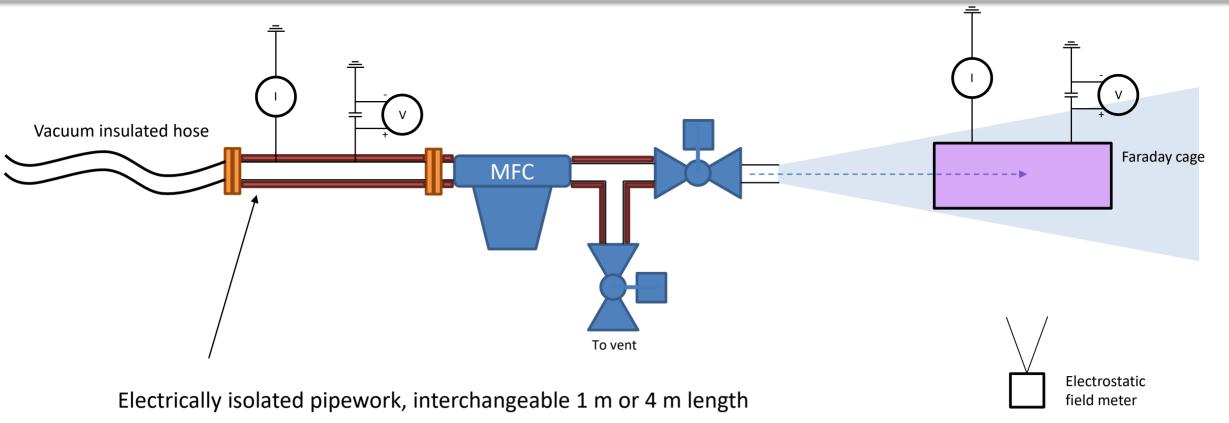
 Electrostatic measurements on pipeline, to estimate the charging current / charge density in bulk liquid flow



To provide base level electrostatic measurements to draw practical implications from – i.e.
 comparison of propensity to ignite vs organics











Instrumentation

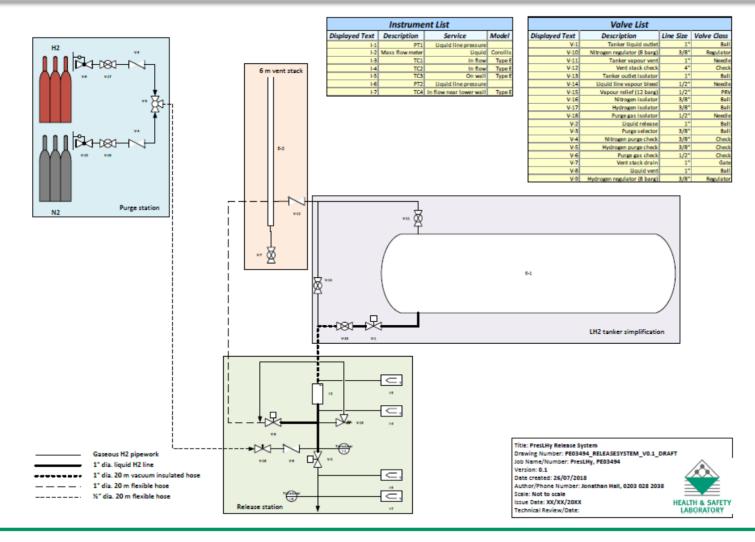
- Electrometer measuring charge on pipework and second on faraday cage
- Electrostatic field meter measuring background charge from plume/jet
- Local humidity from local weather station

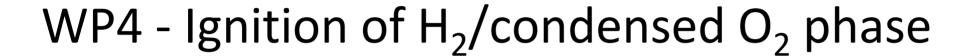
Infrastructure

- 1 m / 4 m long isolated section of pipe
- Shield plate for field meter
- Shielded cable from pipework to control room for electrometer
- Faraday cage, dimensions 0.5 m * 0.5 m * 1.0 m











- Bespoke release conditions/set up.
- Condensed phase will be made up of pure oxygen on first pass, any ignitions may be repeated with a condensed phase made up of air.

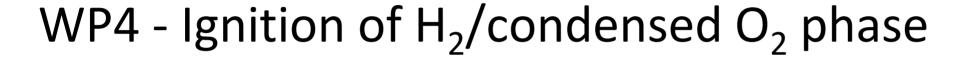
Work Package	Experimental Subtask	Test No.	Experiment Title	Ignition Type
4	4.5	4.5.1	Ignition of H ₂ /condensed O ₂ phase	Explosive detonator
4	4.5	4.5.2	Ignition of H ₂ /condensed O ₂ phase	Electric spark, multiple heights
4	4.5	4.5.3	Ignition of H ₂ /condensed O ₂ phase	Chemical ignitor, >10 kJ
4	4.5	4.5.4	Ignition of H ₂ /condensed O ₂ phase	Mechanical, drop weight
4	4.5	4.5.5	Ignition of H ₂ /condensed air phase	Explosive detonator
4	4.5	4.5.6	Ignition of H ₂ /condensed air phase	Electric spark, multiple heights
4	4.5	4.5.7	Ignition of H ₂ /condensed air phase	Chemical ignitor, >10 kJ
4	4.5	4.5.8	Ignition of H ₂ /condensed air phase	Mechanical, drop weight
4	4.5	4.5.9	Rapid phase transition	Water deluge (fire hose)
4	4.5	4.5.10	Rapid phase transition	Water deluge (sprinkler)

WP5 - Combustion with congestion/confinement



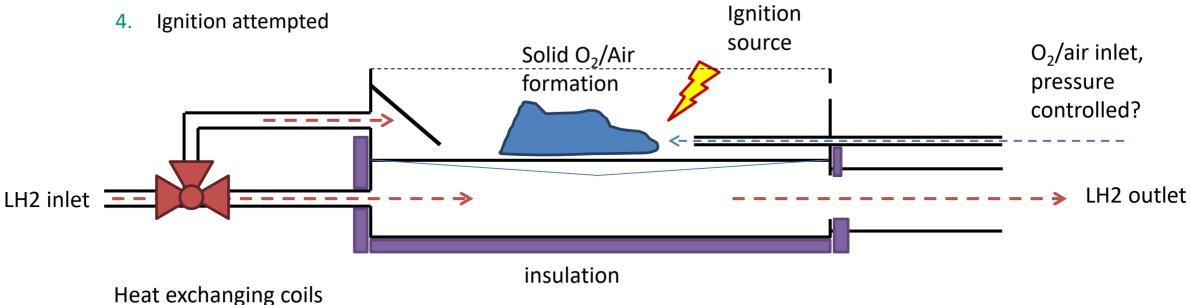
- Distance arcs of 215 mand 475 m
- Assuming 135 dB limit at extremes
- Equates to:
 - 0.137 kg TNT @ 215 m
 - 1.480 kg TNT @ 475 m







- Bespoke release conditions/set up
 - 1. LH2 fed into lower chamber until upper plate is cold
 - 2. O2/air fed onto top surface of plate to form solid deposition
 - 3. LH2 feed then turned onto top surface and deposition

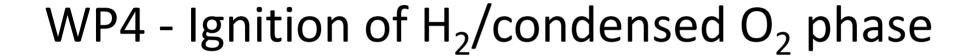


Rapid phase transition tests could use same rig or simply spills onto the ground for longer





- Used sample equivalent to 0.11 kg of TNT
- Previous NASA work used 0.024 kg of O₂ with LH₂ in excess, LH₂
 quantity did not affect ignition sensitivity
- Used PU insulated PE containers
- Explosive yield = $2 \text{ kg TNT per } 1 \text{ kg } LH_2 + O_2$
- Therefore maximum of 0.15 kg of TNT = 0.075 kg $LH_2 + O_2$



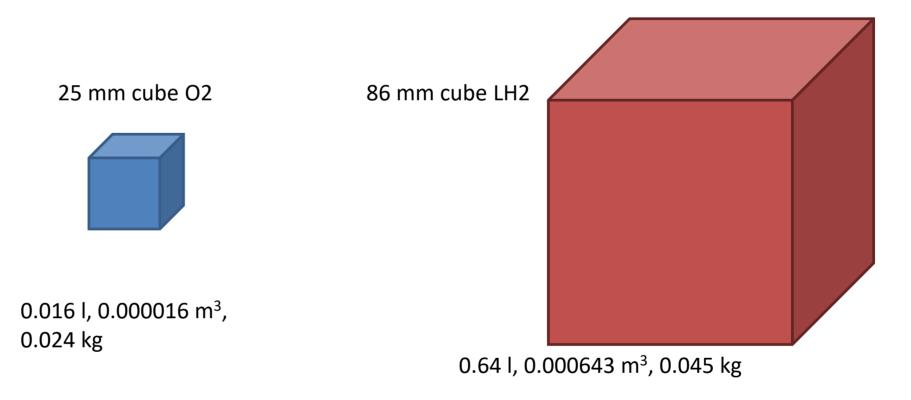


- The density of solid oxygen ranges from 21 cm³/mol in the α-phase, to 23.5 cm³/mol in the γ-phase. (Roder, H. M. (1978). "The molar volume (density) of solid oxygen in equilibrium with vapor" (reprint). Journal of Physical and Chemical Reference Data. 7 (3): 949.)
- Therefore: $21 \text{ cm}^3/\text{mol} = 0.0476 \text{ mol/cm}^3$
- O_2 molar mass is 32 g/mol
- Density of solid $O_2 = 32*0.0476 = 1.524 \text{ g/cm}^3 = 1524 \text{ kg/m}^3$
- If 0.024 kg of solid O_2 is generated that takes up 0.016 l of space
- If the remainder of 0.075 kg of mixture is LH², that leaves 0.045 kg of LH² (density 70 kg/m³) which takes up 6.43e⁻⁴ m³ or 0.64 l

WP4 - Ignition of H₂/condensed O₂ phase

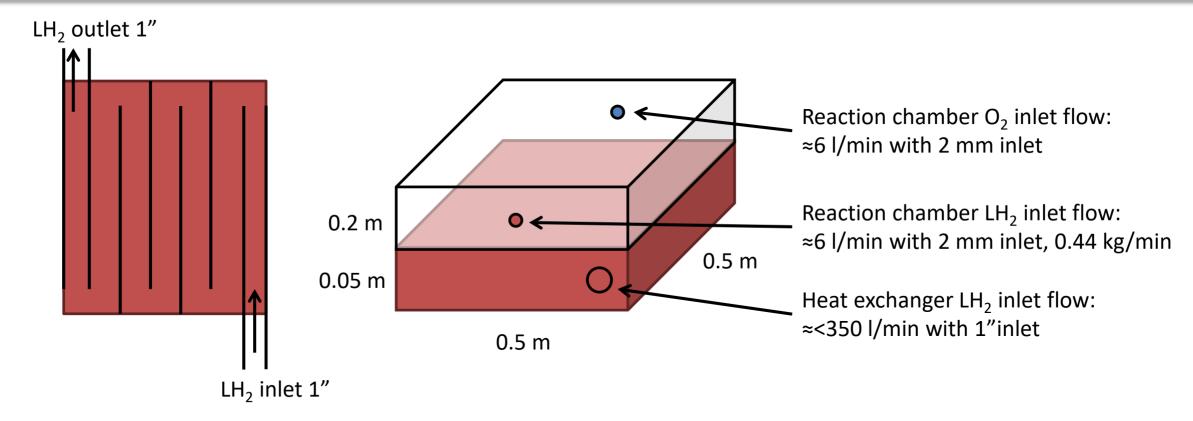


- If 0.5×0.5 m tray is used, 0_2 level = 0.064 mm, LH₂ level = 2.6 mm
- If 0.25×0.25 m tray is used, 0_2 level = 0.26 mm, LH_2 level = 10.3 mm



WP4 - Ignition of H₂/condensed O₂ phase



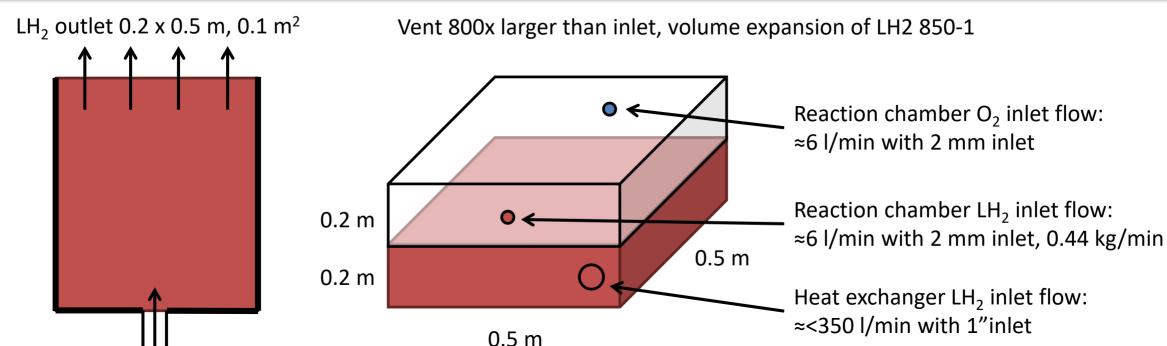


Reaction chamber LH2 inlet sized to give maximum allowable quantity in \approx 30 secs. Total coverage of solid O₂ is required for most ignition sensitive mixture therefore surface area of rig adjusted to suit.

WP4 - Ignition of H₂/condensed O₂ phase

LH₂ inlet ½", 1.27e-4 m²



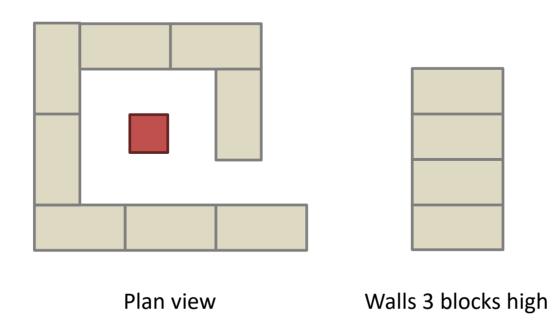


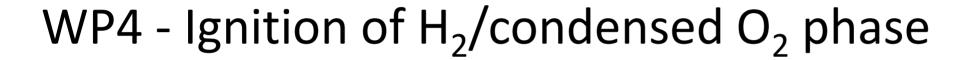
Reaction chamber LH2 inlet sized to give maximum allowable quantity in \approx 30 secs. Total coverage of solid O₂ is required for most ignition sensitive mixture therefore surface area of rig adjusted to suit.





- Surround reaction chamber with concrete blocks
- Blast pressure measurement within the concrete walls







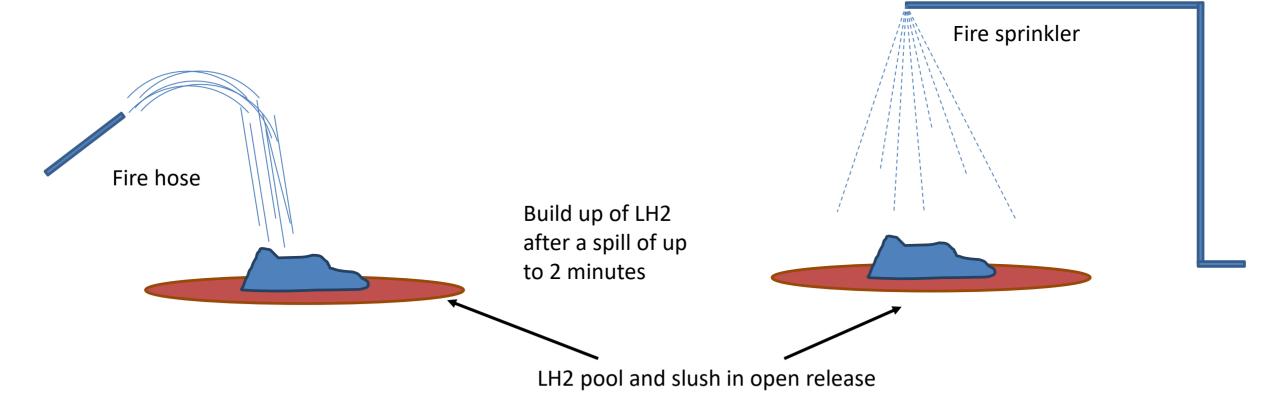
Energy of ignition of LH2/Condensed O₂ / N₂ based on an energy of impact source, Litchfield and Perlee:

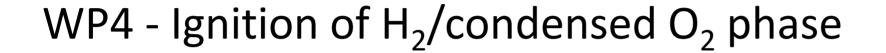
	Percent N2 in	Initiation		Failure		
	Condensate	Velocity m/sec	Energy (kJ)	Velocity m/sec	Energy (kJ)	
Pure O ₂	0	506	1.80	506	1.80	
	23	695	3.39	594	2.48	
	47	1091	8.37	991	6.90	
Air	78	-	-	1487	15.55	
			↑		1	
		Ini	tiated at	Did not initiate		
		energies at or		at energies		
		ab	ove	below		





Tests 4.5.9 and 4.5.10







Instrumentation

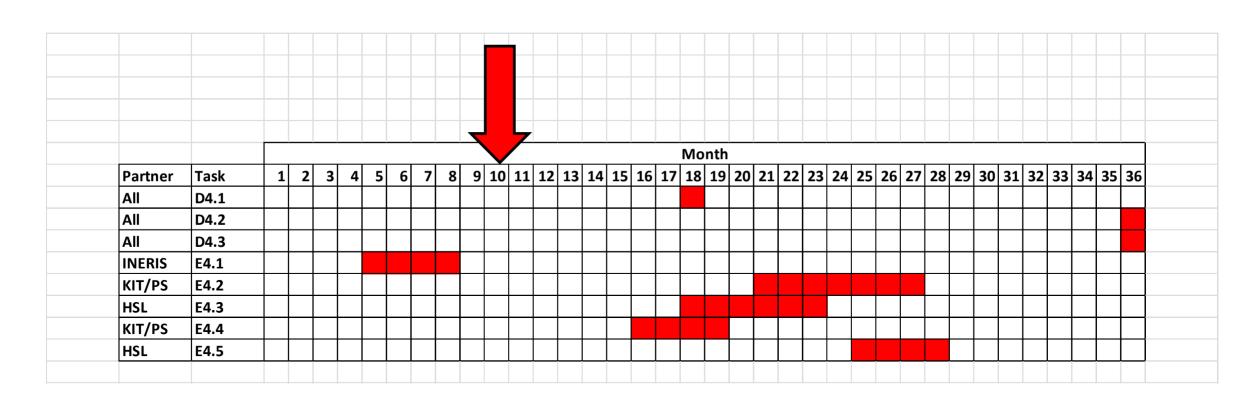
- 2x blast pressure transducers within concrete enclosure walls (ranged 0-5 bara, 500 kHz logging rate)
- Audible sound meters at 50 m and 100 m approximately
- Remote ignition system with multiple outputs
- Thermocouples within reaction chamber/rig
- High speed video and IR

Infrastructure

- Reaction chamber/rig
- Protective concrete block wall around chamber
- Protective steel shield in front of release station 1 m wide x 2 m high











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