

# HySafe Research Priorities Workshop

Washington DC, 2014-11-(10-11)

## Passive ventilation of enclosures with one vent, the uniformity criterion, and validation of pressure peaking phenomenon (unignited releases)

Shentsov Volodymyr, Molkov Vladimir

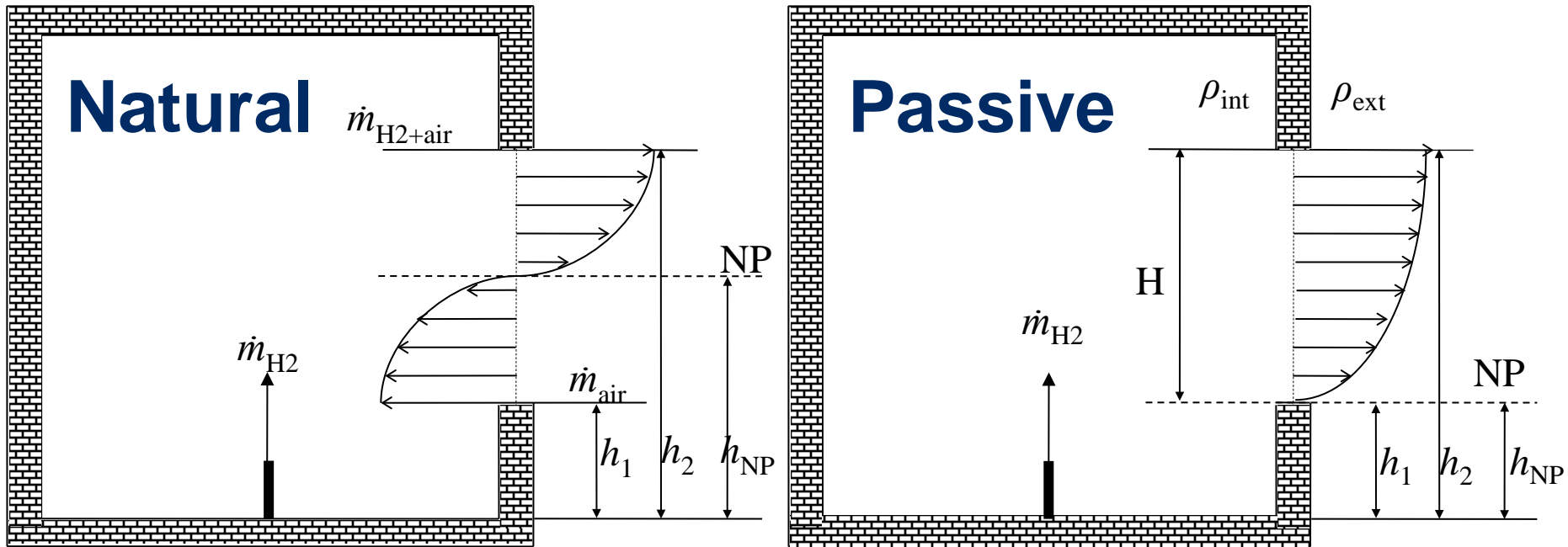
Hydrogen Safety Engineering and Research Centre (HySAFER)

<http://hysafer.ulster.ac.uk/>

# Outline

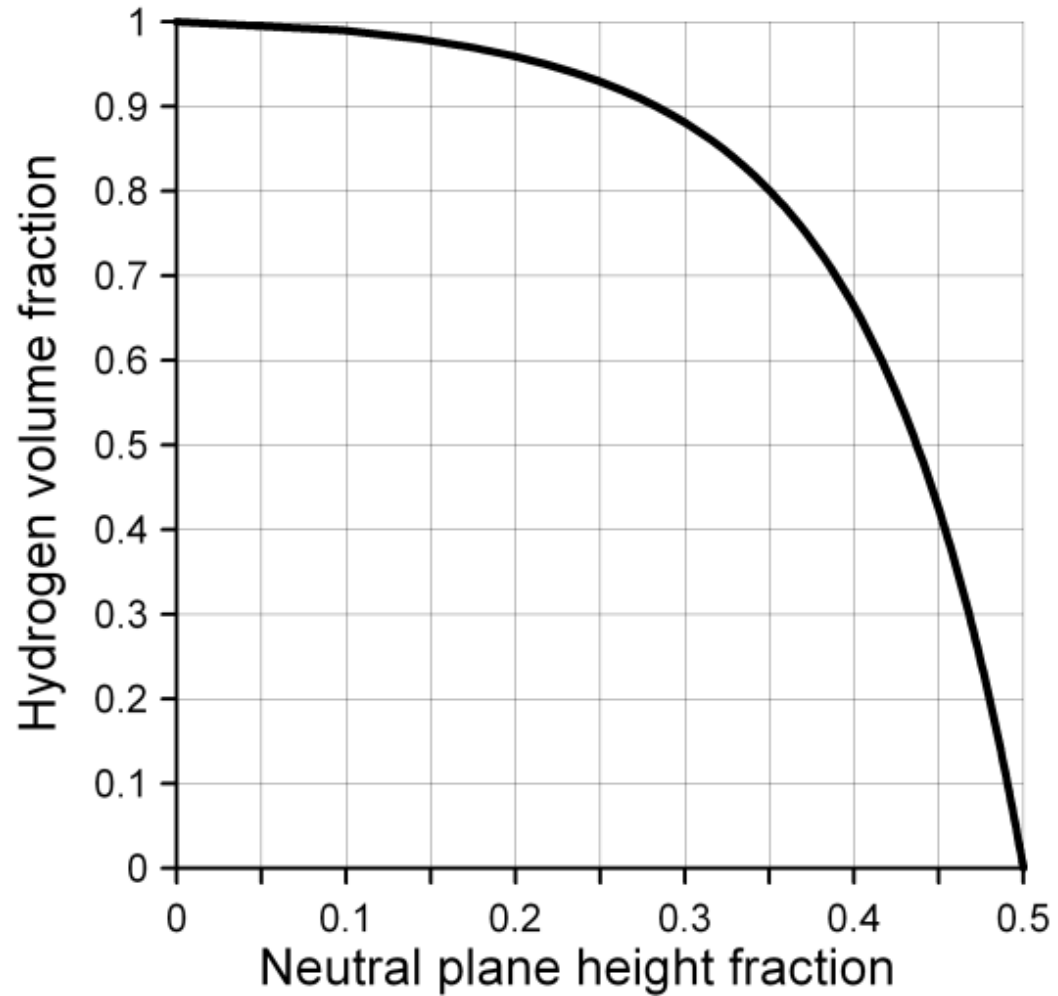
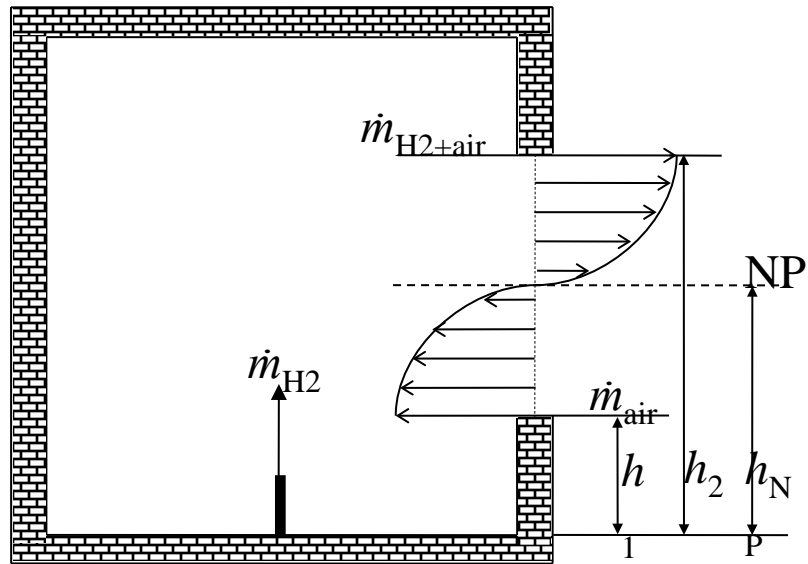
- ❖ Natural and passive ventilation
- ❖ The model and the nomogram for steady state hydrogen concentration in an enclosure with one vent (sustained leak)
- ❖ The criterion for mixture uniformity during leak in an enclosure with one vent
- ❖ The model and the nomogram for leak flow rate limit leading to 100% hydrogen accumulation in an enclosure with time
- ❖ Validation of pressure peaking phenomenon

# Ventilation



- ❖ **Natural ventilation** equations for air ventilation in buildings are derived in the assumption of equality of flow in and out of the enclosure (neutral plane is at the half of a vent height).
- ❖ **Passive ventilation:** neutral plane for lighter than air gases can be anywhere below the half of a vent height.

# Neutral plane location



# Mole fraction $X$ : natural vs passive

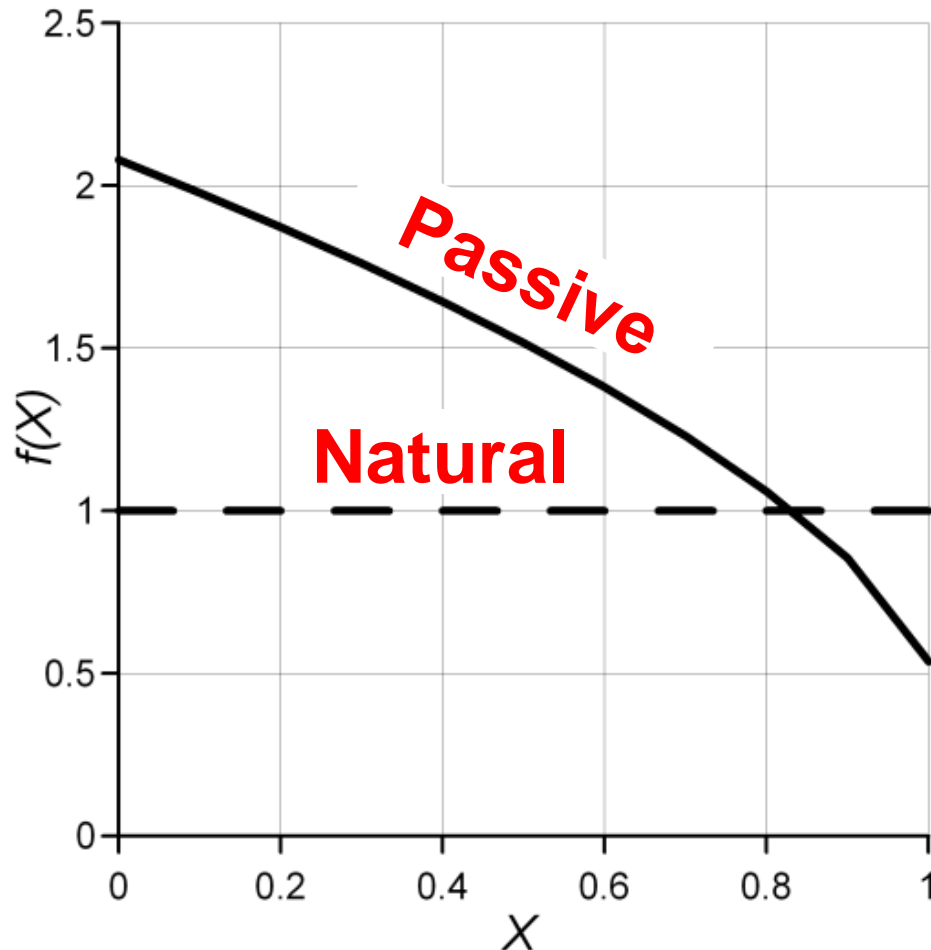
Natural ventilation: 
$$X = \left[ \frac{Q_0}{C_D A (g' H)^{1/2}} \right]^{2/3}$$

Passive ventilation: 
$$X = f(X) \cdot \left[ \frac{Q_0}{C_D A (g' H)^{1/2}} \right]^{2/3}$$

Difference: 
$$f(X) = \left( \frac{9}{8} \right)^{1/3} \cdot \left\{ \left[ 1 - X \left( 1 - \frac{\rho_{H_2}}{\rho_{air}} \right) \right]^{1/3} + (1 - X)^{2/3} \right\}$$

# Safety design: natural or passive?

Difference:  $f(X) = \left(\frac{9}{8}\right)^{1/3} \cdot \left\{ \left[ 1 - X \left( 1 - \frac{\rho_{H_2}}{\rho_{air}} \right) \right]^{1/3} + (1 - X)^{2/3} \right\}$



**Natural ventilation equation (compared to passive):**

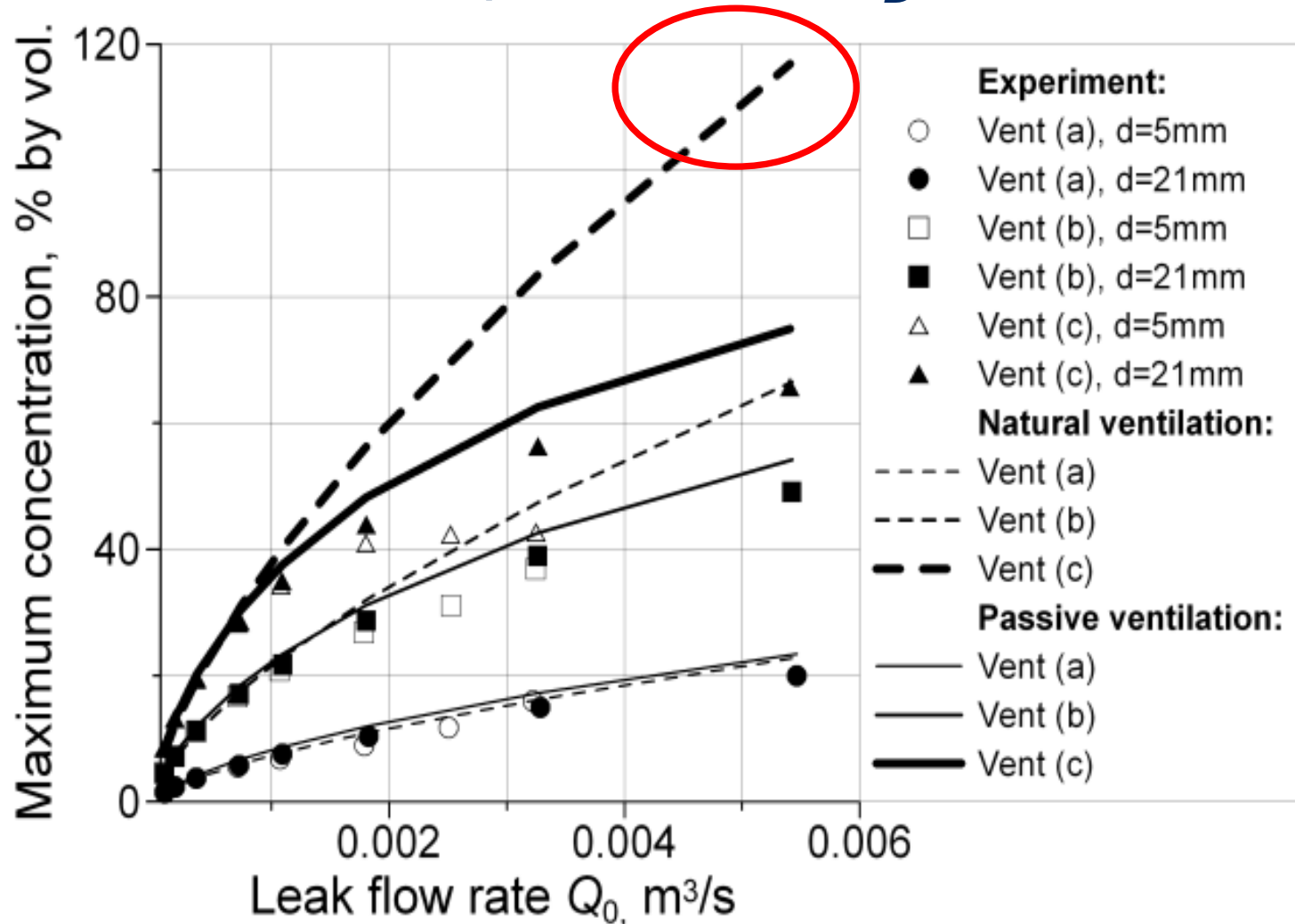
- ❖ Underestimate **x2 (lean)**
- ❖ Overestimate **x2 (rich)**

# Validation tests with helium (CEA)

- ❖ Experiments were carried out by CEA (France) in the enclosure with sizes  $H \times W \times D = 1.26 \times 0.93 \times 0.93$  m.
- ❖ One vent located on a wall near the ceiling.
- ❖ Three different vent sizes were studied:
  - Vent (a)  $W \times H = 90 \times 18$  cm,
  - Vent (b)  $18 \times 18$  cm,
  - Vent (c)  $90 \times 3.5$  cm.
- ❖ Release of helium was directed upward from a tube located centrally 21 cm above the floor with internal diameter either 5 mm or 21 mm.
- ❖ Release rates ranges  $5 \div 300$  NI/min.

# Predictions vs experiment $C_{max}$

Natural ventilation with “tuned”  $C_D=0.25$  - dashed lines.  
Passive ventilation equation with  $C_D=0.60$  solid lines.





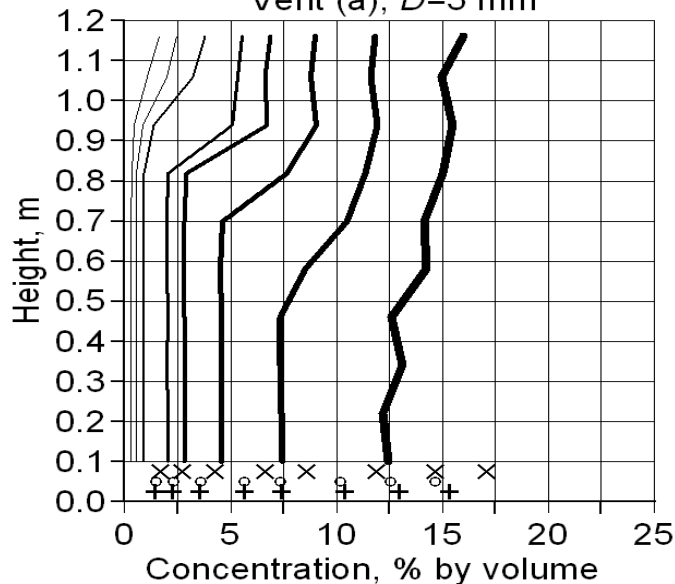
# Experiments

Vent (a):  $H \times W = 18 \times 90$  cm;

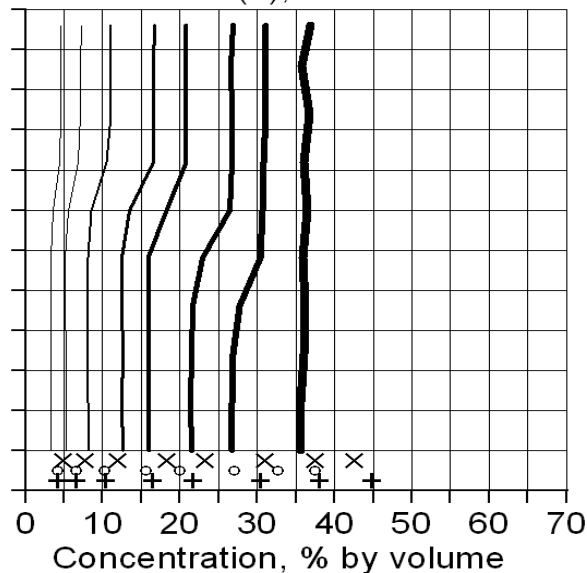
Vent (b):  $18 \times 18$  cm;

Vent (c):  $3.5 \times 90$  cm;

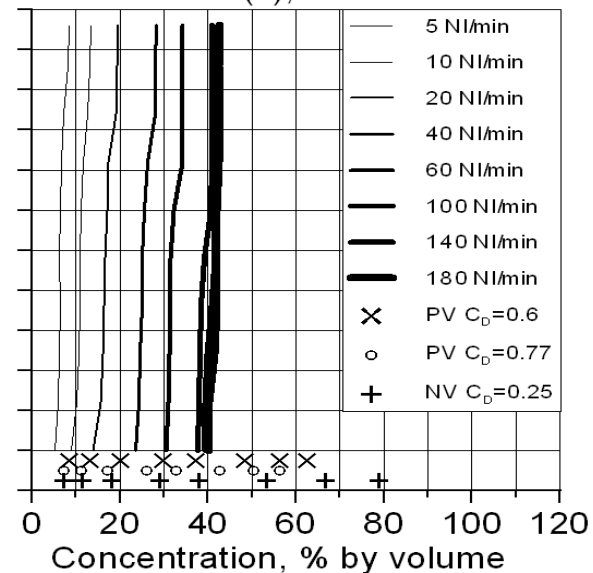
Vent (a),  $D=5$  mm



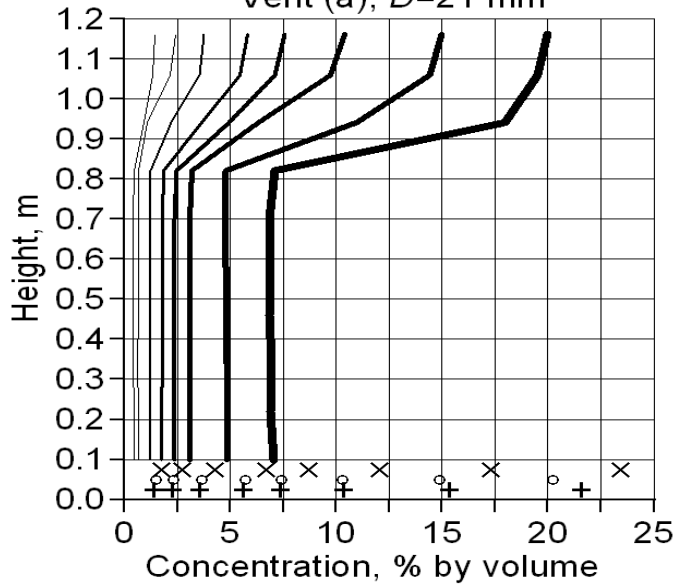
Vent (b),  $D=5$  mm



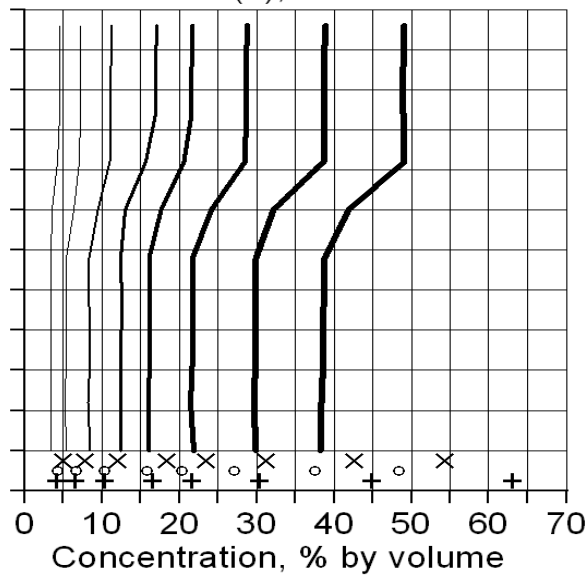
Vent (c),  $D=5$  mm



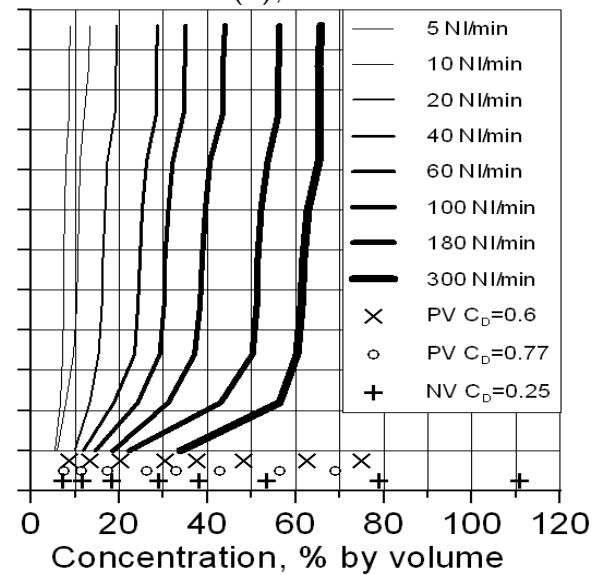
Vent (a),  $D=21$  mm



Vent (b),  $D=21$  mm



Vent (c),  $D=21$  mm

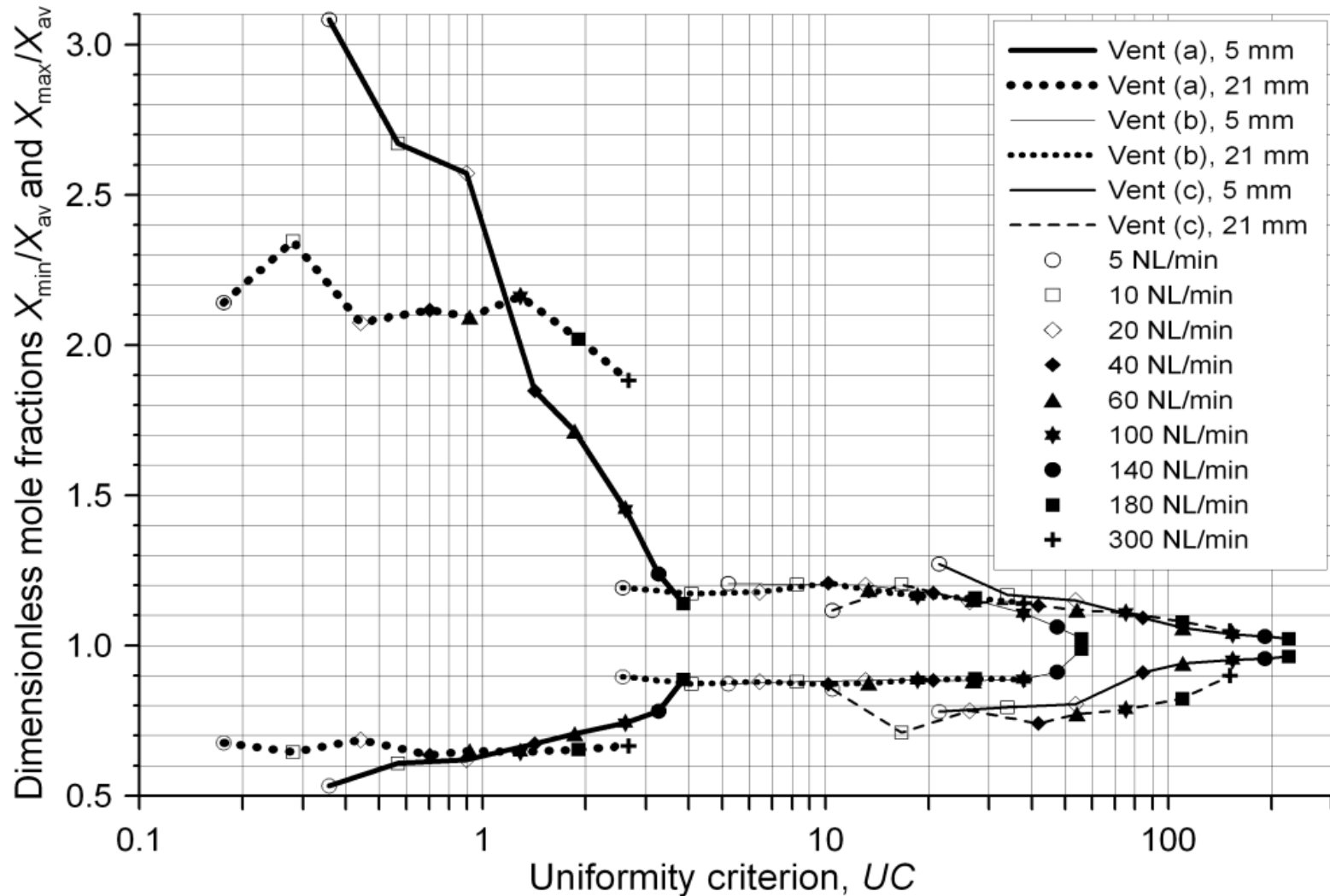


# The criterion of uniformity

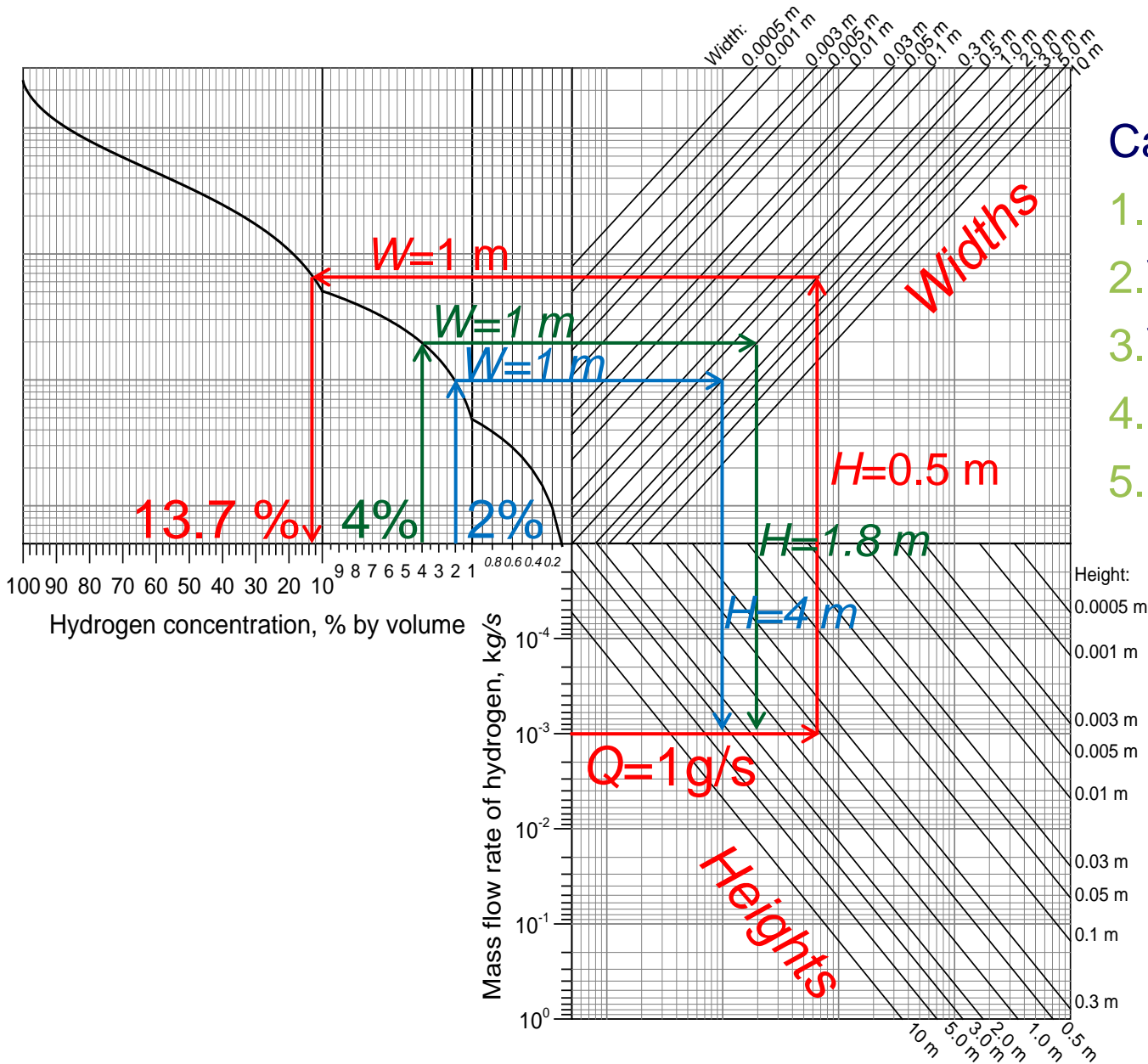
$$UC = \frac{V^{2/3} \sqrt{D} \dot{m}_{ent}(x)}{A \sqrt{H} \dot{m}_{mix}}$$

$$\dot{m}_{ent}(x) = K_1 M_0^{1/2} \rho_{mix}^{1/2} x$$

$$\dot{m}_{mix} = \dot{m}_{H_2} + \dot{m}_{air}$$



# Nomogram (concentration $C_{max}$ )

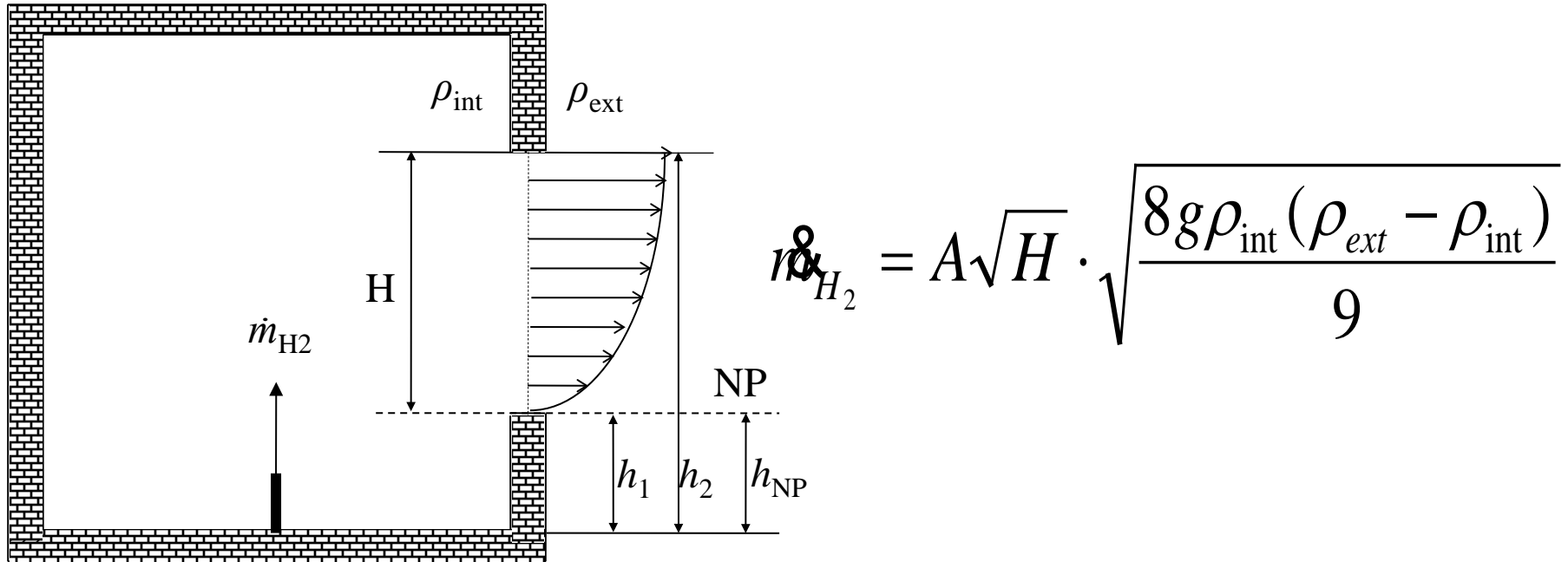


Calculation example:

1. Release rate (1 g/s)
2. Vent Height (0.5 m)
3. Vent width (1 m)
4. Function curve
5. Concentration (13.7%)

# Release limit (100% H2 accumulation)

Depending on the release rate and vent size there can be 100% of hydrogen accumulation in the enclosure.



Equation gives the lower limit of release flow rate that leads to 100% of hydrogen accumulation in enclosure.

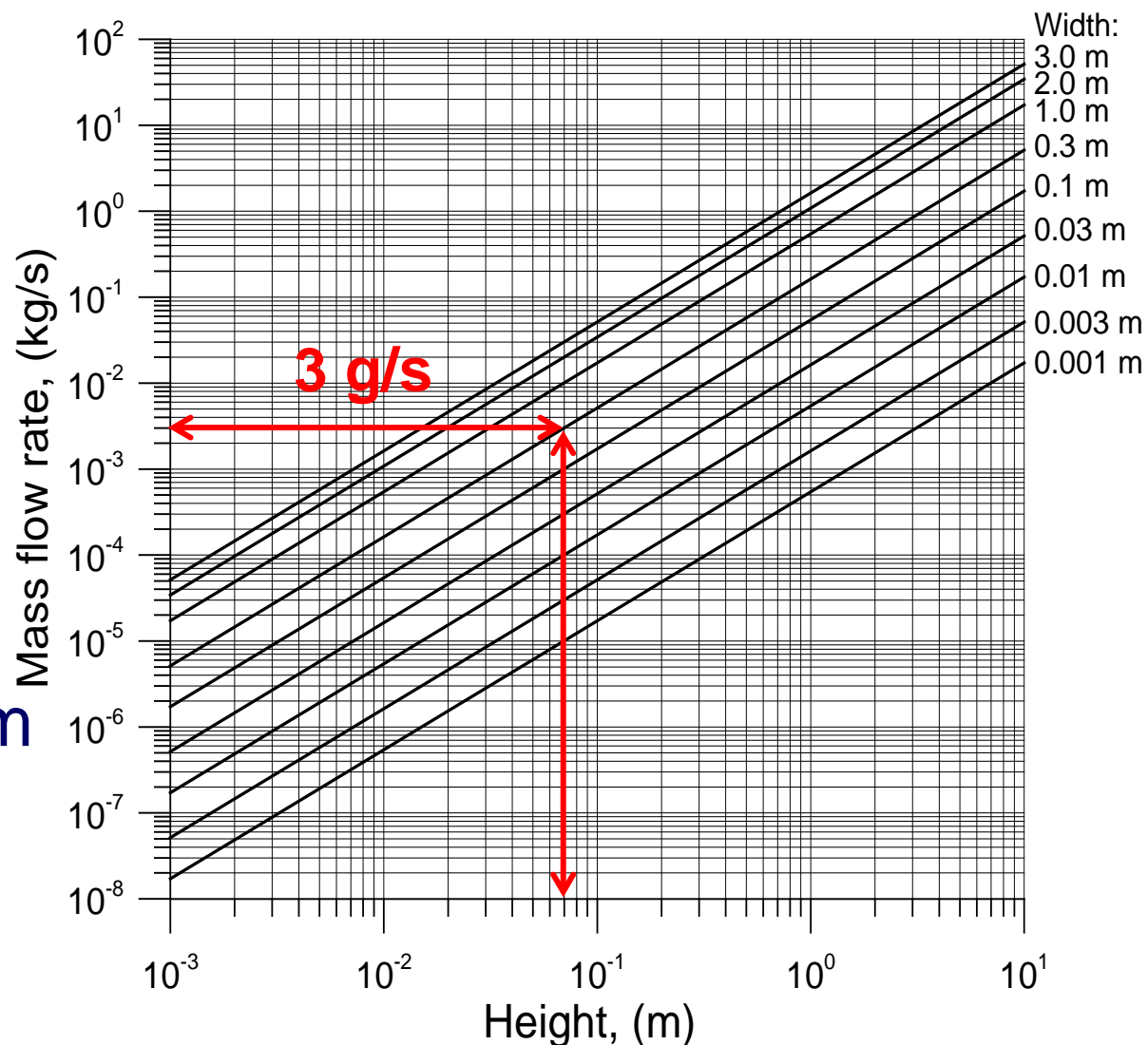
**(This limit has to be exceeded to have the pressure peaking phenomenon – see in following slides)**

# Nomogram: 100% accumulation

Example of use: sustained release of hydrogen 3 g/s in enclosure with a vent of  $W=0.3$  m and  $H=0.07$  m will lead to 100% accumulation.

No dependence on enclosure volume (only accumulation time will depend).

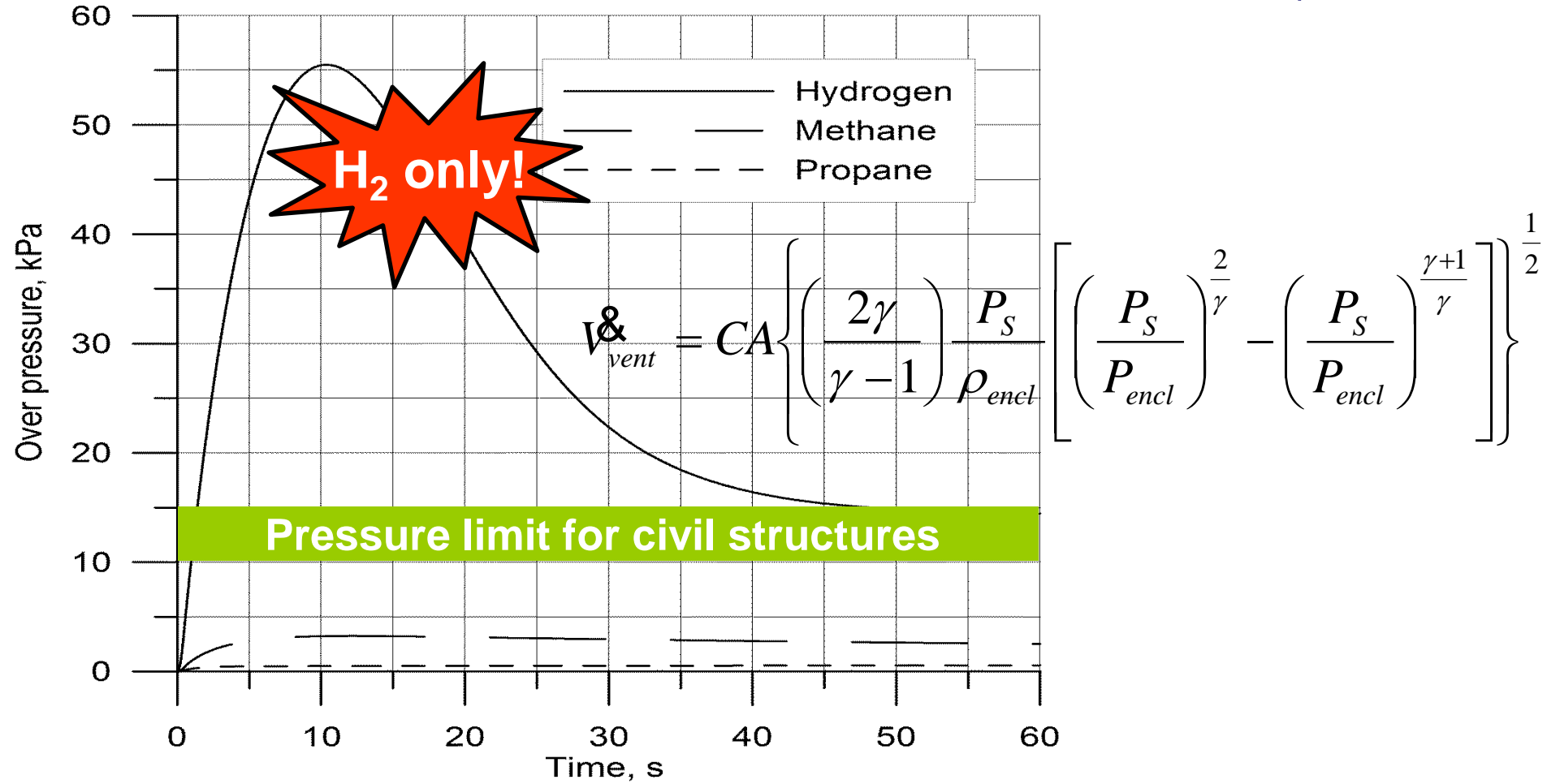
Reproduces previous nomogram at  $C_{max}=100\%$



# Pressure peaking phenomena

Small garage  $L \times W \times H = 4.5 \times 2.6 \times 2.6$  m (“brick” vent).

Mass flow rate 390 g/s (350 bar, 5.08 mm orifice).

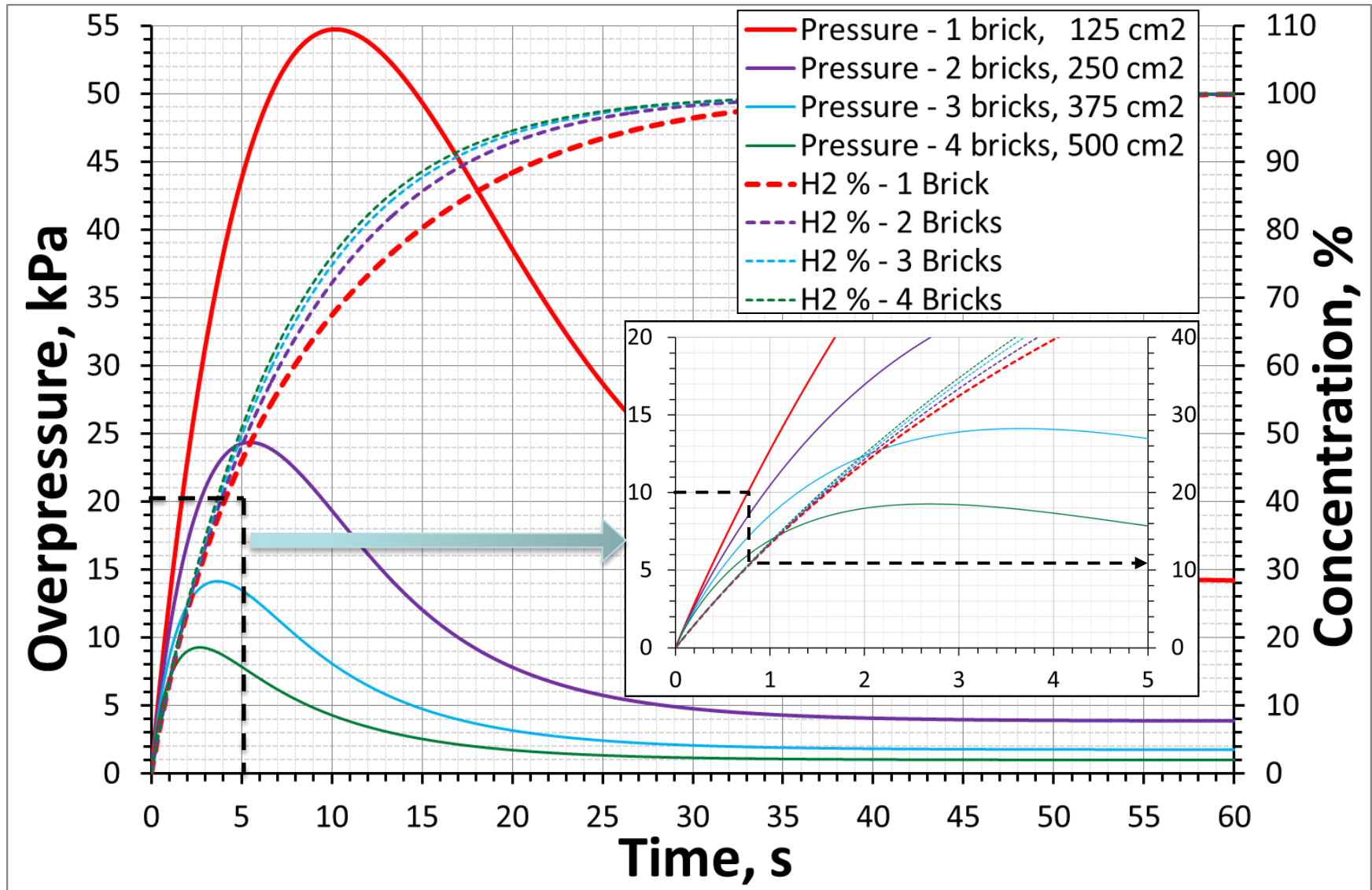


**Solution: decrease PRD orifice size (and at the same time increase fire resistance of onboard storage)**

# Pressure peaking phenomena

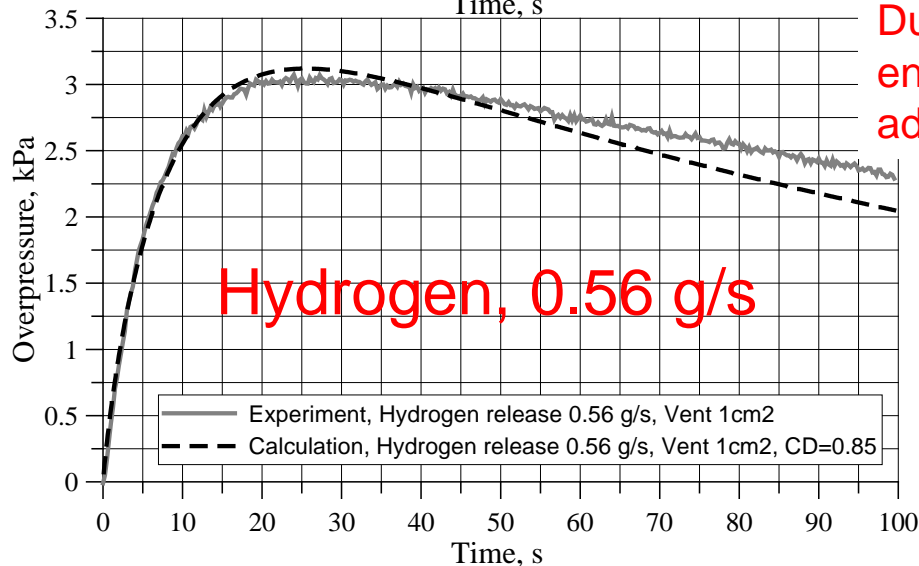
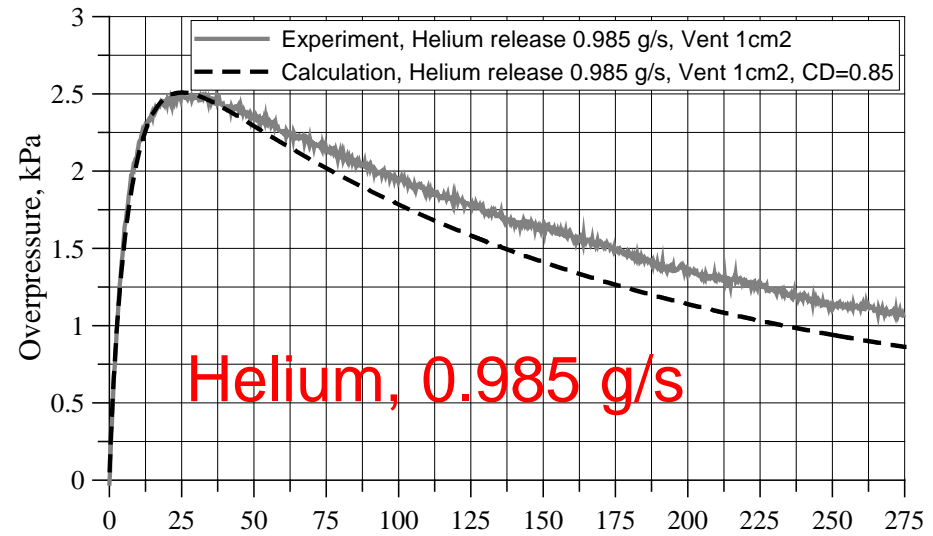
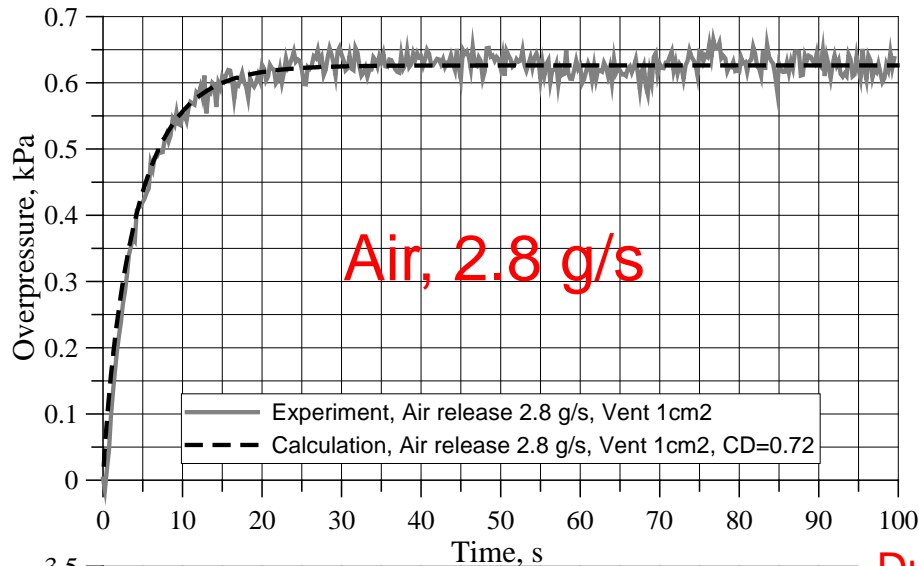
Small garage  $L \times W \times H = 4.5 \times 2.6 \times 2.6$  m (“brick” vent).

Mass flow rate 390 g/s (350 bar, 5.08 mm orifice).

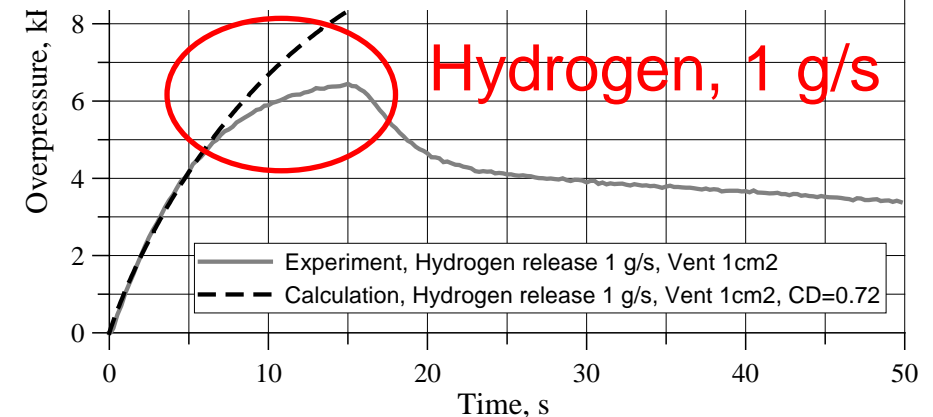


# Pressure peaking: validation

Formula for the pressure peaking calculation had been validated against experimental result obtained by KIT for the steady state release.



Due to higher pressure the experimental enclosure starts to “breathe” resulting in additional opening





# Conclusions

- ❖ The analytical model for uniform concentration in an enclosure with one passive vent for sustained leak is developed, validated, and used to build engineering nomograms.
- ❖ The criterion for mixture uniformity in a vented enclosure is suggested.
- ❖ The nomogram to calculate the mass flow rate limit leading to 100% gas concentration in the enclosure as a function of the vent width and height is developed (flow rate should be above this limit to apply the pressure peaking phenomenon calculations).
- ❖ Experiments with release of air, helium and hydrogen performed by KIT proved the existence of pressure peaking phenomena (PPP).
- ❖ It can be concluded that the PPP is essential part of hydrogen safety engineering (HSE) for all indoor use of HFC systems.
- ❖ It is recommended that value  $CD=0.6$  is accepted for HSE as a conservative one.

**Learn more!**



**THANK YOU**

## Acknowledgements:

- ❖ EPSRC SUPERGEN HFC HUB (<http://www.h2fcsupergen.com/>)
- ❖ FCH JU HyIndoor project (<http://www.hyindoor.eu/>)

**H<sub>2</sub>FCSUPERGEN**  
THE HYDROGEN AND FUEL CELL RESEARCH HUB

 Hyindoor

**MSc in Hydrogen Safety Engineering (distance learning course):**

<http://www.ulster.ac.uk/elearning/programmes/view/course/10139>

**Fundamentals of Hydrogen Safety Engineering**

(free download eBook, [www.bookboon.com](http://www.bookboon.com), 2012)